

Back-of-the-Envelope Physics

Winter Term 2023/24

Sheet 10

1. Show that Maxwell's equations and the Lorentz-force equation of motion are invariant under parity P and time reversal T . In the process, determine the transformation properties of the fields \vec{E} , \vec{B} under P and T .

2. a) Show that the convective derivative of the velocity field $\vec{v}(t, \vec{x})$ can be written as

$$D_t \vec{v} \equiv \partial_t \vec{v} + (\vec{v} \cdot \vec{\partial}) \vec{v} = \partial_t \vec{v} + \vec{\Omega} \times \vec{v} + \frac{1}{2} \vec{\partial} v^2 \quad (1)$$

where $\vec{\Omega} \equiv \vec{\partial} \times \vec{v}$ is the vorticity field.

b) Show that Euler's equation in a gravitational potential ϕ , $D_t \vec{v} = -(\vec{\partial} P)/\rho - \vec{\partial} \phi$, implies Bernoulli's theorem. This states that in a stationary flow ($\partial_t \vec{v} = 0$)

$$\frac{P}{\rho} + \frac{1}{2} v^2 + \phi = \text{const} \quad (2)$$

along a streamline. Also, (2) holds everywhere if the stationary flow is irrotational ($\vec{\Omega} \equiv 0$).

3. Consider a cylindrical tube with inner radius a and length $l \gg a$. The z -axis is chosen to coincide with the symmetry axis of the cylinder. An incompressible fluid with dynamical viscosity μ is flowing through the tube in the z -direction. It is driven by a pressure difference ΔP between the two ends of the tube, resulting in a homogeneous pressure gradient $-\Delta P/l$ in the z -direction inside the tube. Assume that the flow has reached a steady state.

- a) Using dimensional analysis, estimate the volume of the fluid per time, \dot{V} , flowing through the cross section of the tube.
- b) Compute the radial velocity profile $v_z(r)$ inside the tube from the Navier-Stokes equation. Assume the boundary condition $v_z(a) = 0$.
- c) Integrate the result of b) to obtain the exact result for \dot{V} .

4. Compute the speed of sound c_s for a gas with equation of state $P(\rho) = \text{const} \cdot \rho^\kappa$. What is κ for air, assuming adiabatic compression? Also derive the temperature dependence of c_s using the ideal gas law.