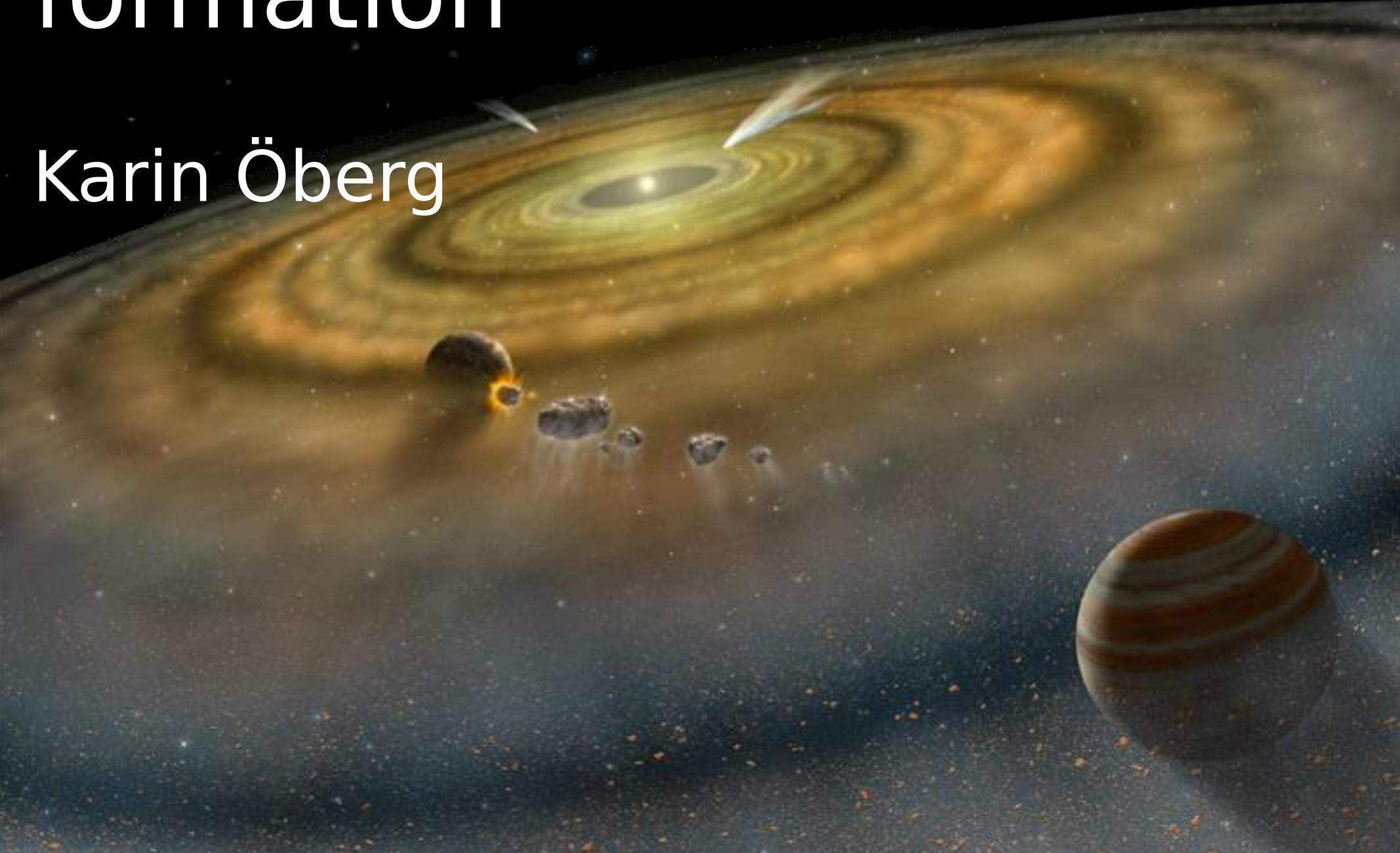


# The chemistry of planet formation

Karin Öberg



# Acknowledgements



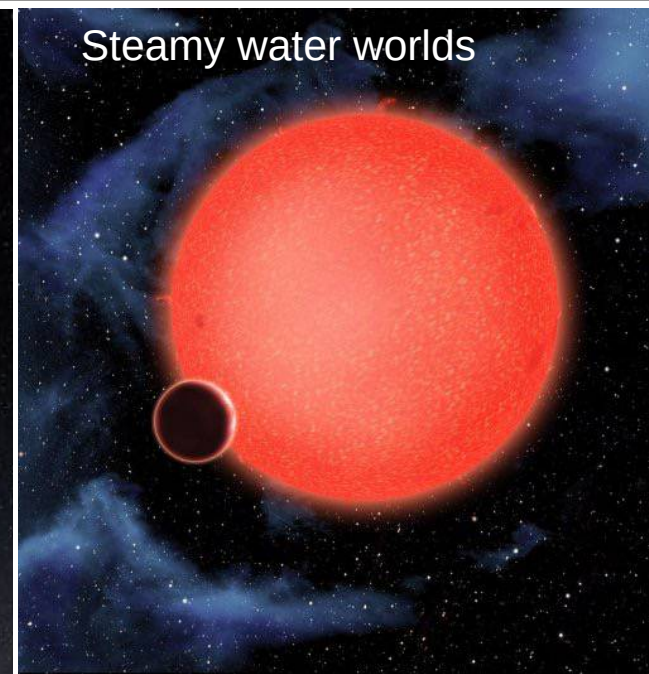
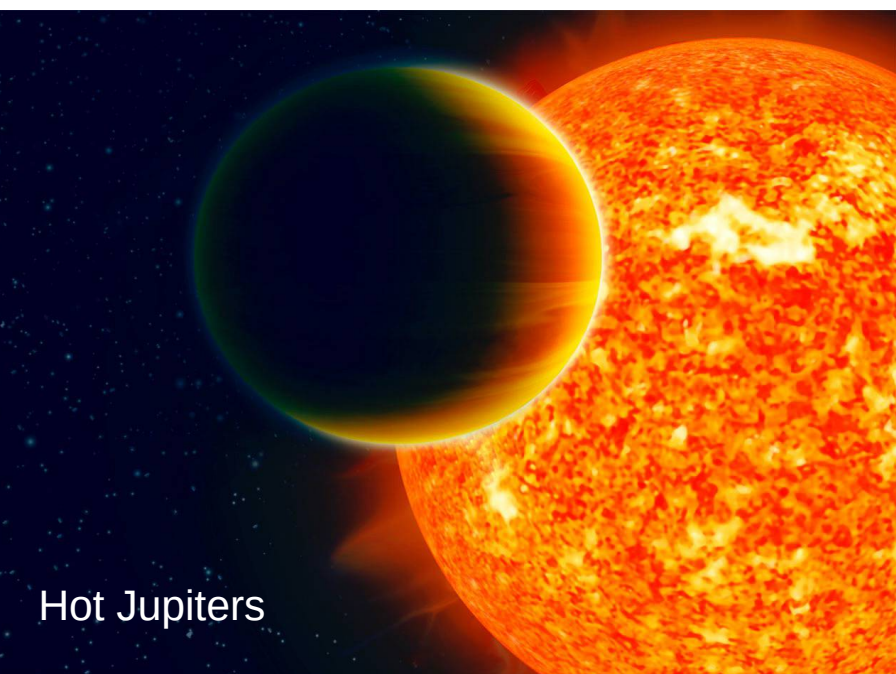
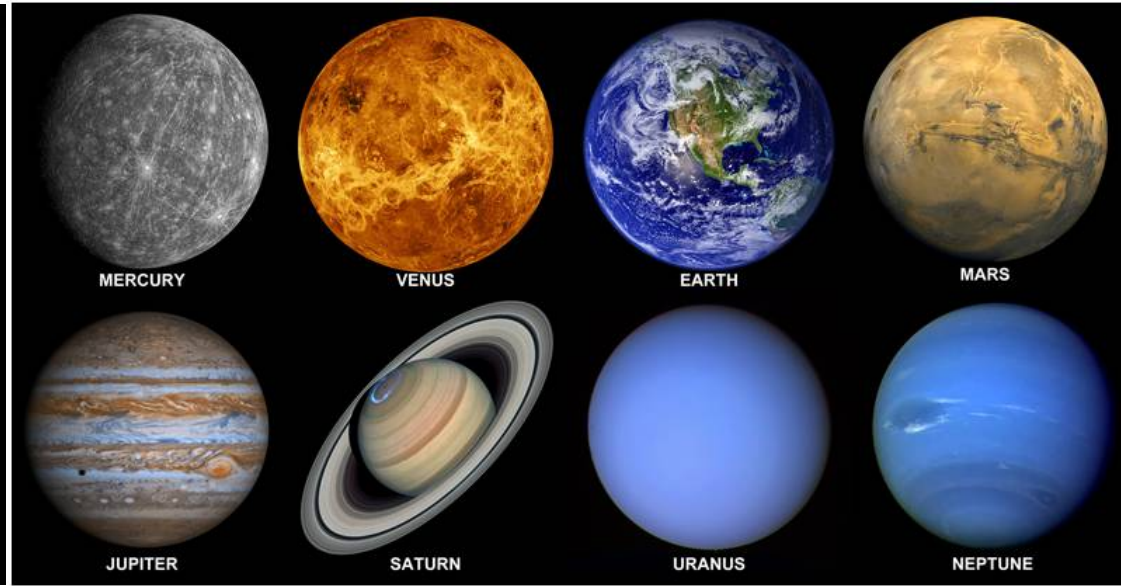
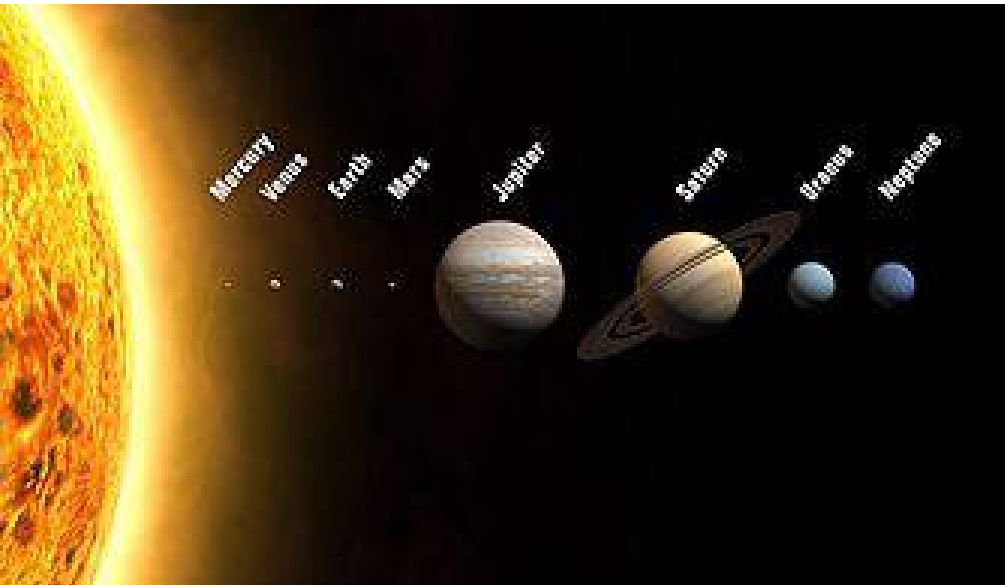
SIMONS  
FOUNDATION



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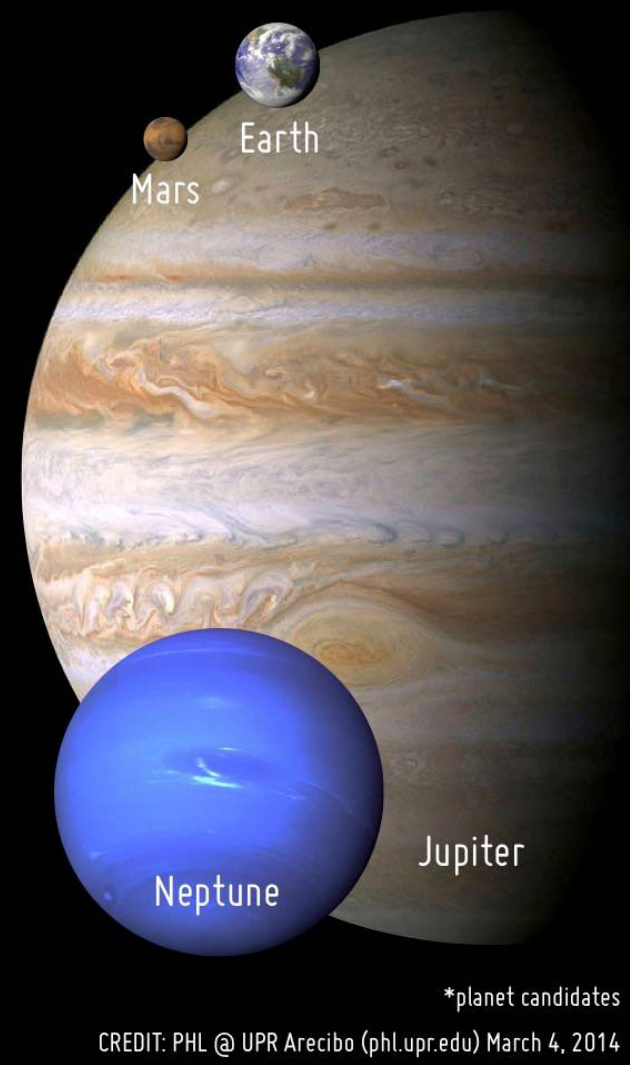
the David &  
Lucile Packard  
FOUNDATION

# A diversity of planet compositions



# Current Potentially Habitable Exoplanets

Ranked in Order of Similarity to Earth

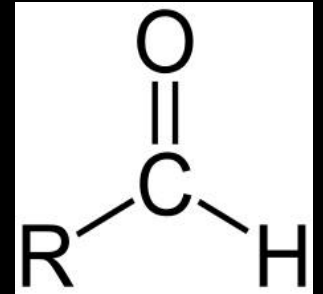
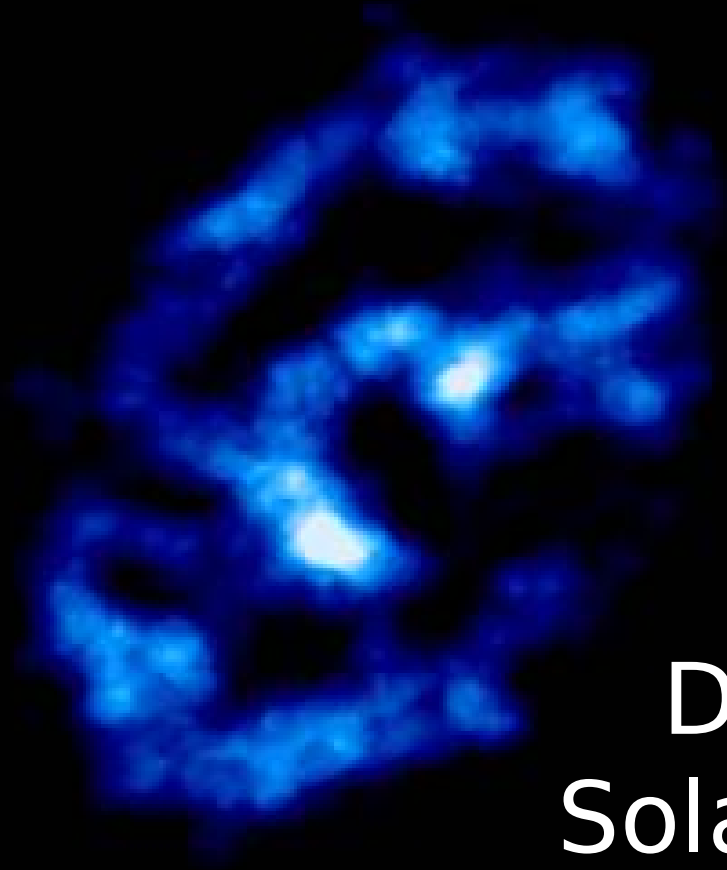


\*planet candidates

CREDIT: PHL @ UPR Arcibo (phl.upr.edu) March 4, 2014

Habitability beyond rocky surfaces and liquid water temperatures... Chemical habitability: access to water, reactive organic molecules... How common is access to the ingredients of life on 'habitable planets'?

# Isotopic labeling

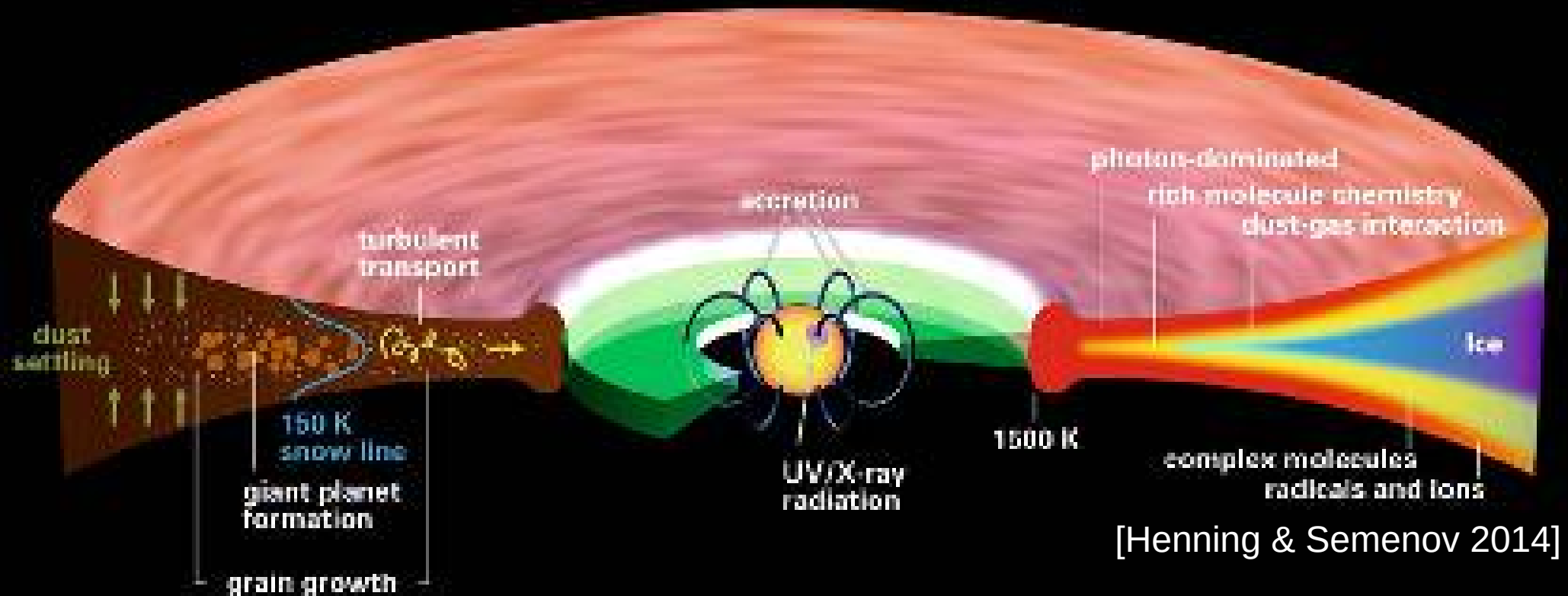


Aldehyde Ion

DCO<sup>+</sup> in a  
Solar nebula  
analog

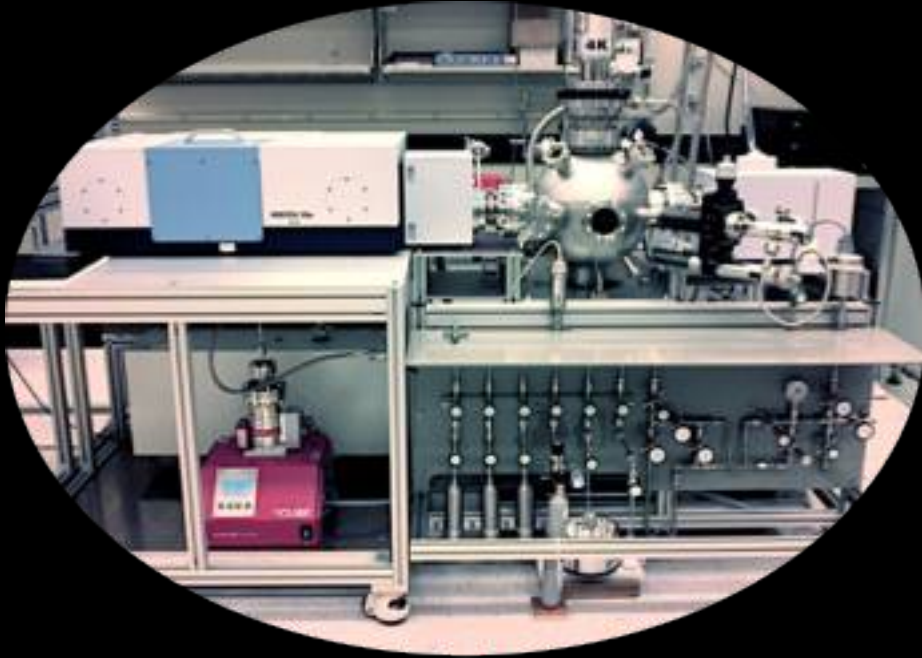
# Protoplanetary disks

are characterized by radial and vertical temperature gradients, grain and gas dynamics and an evolving chemistry



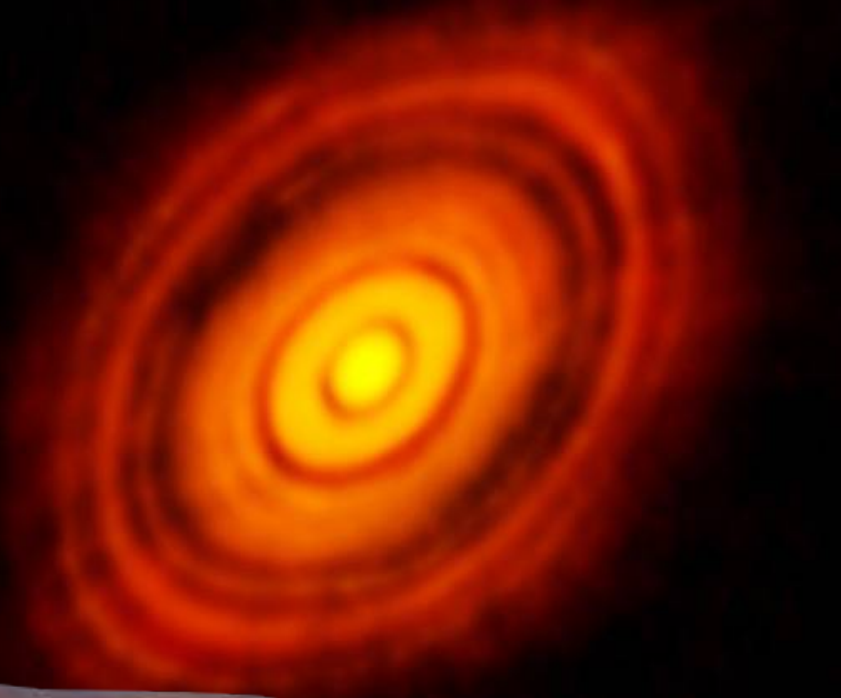
Proposition: the size, elemental composition (e.g. O/H and C/O) and chemical habitability of a nascent planet depends on the chemical composition of the disk material it forms from!

# Characterizing the chemistry of planet formation

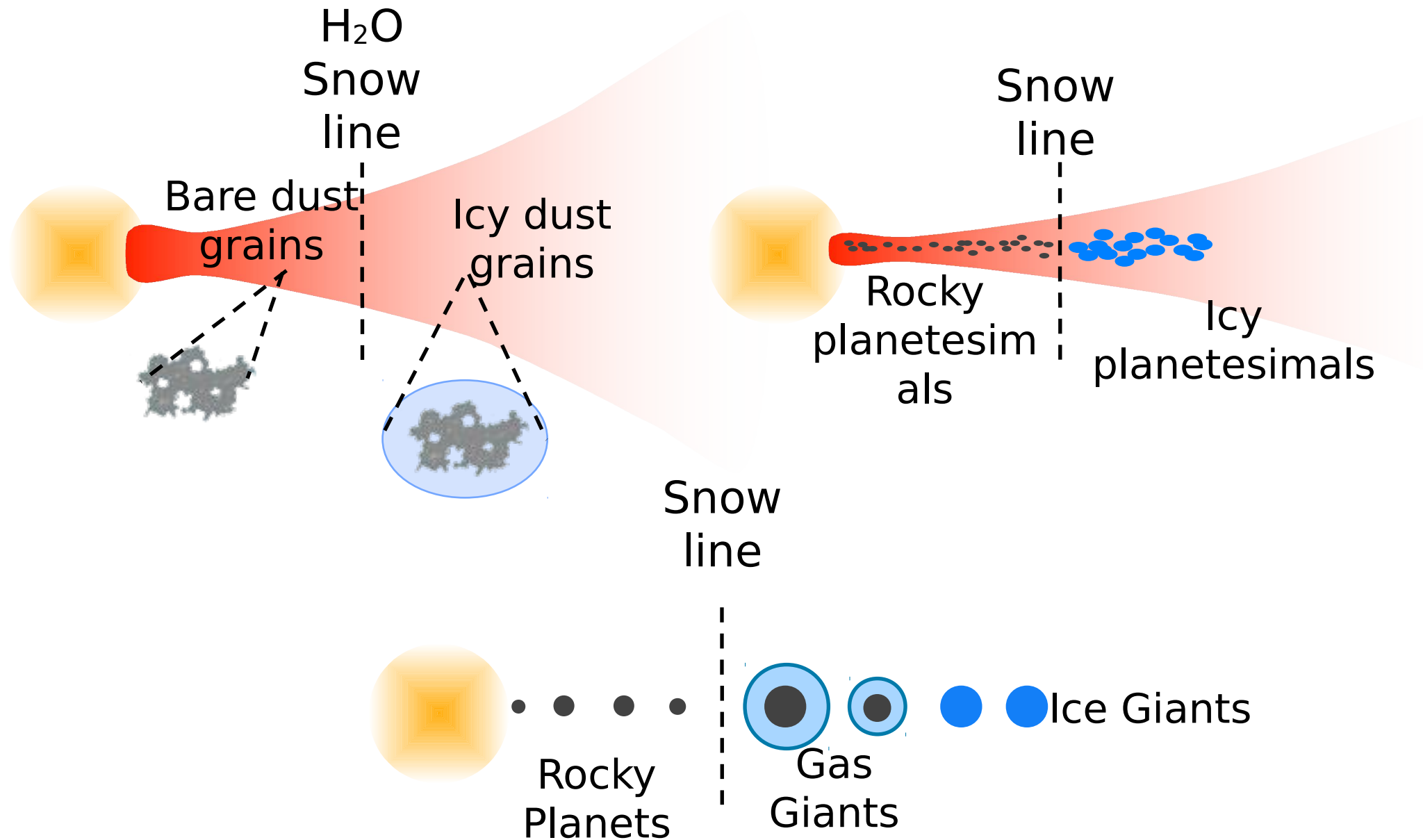


Direct observations of the chemical/molecular compositions of protoplanetary disks and related astronomical objects

Laboratory simulations of the chemical processes that set the disk compositions at different stages of planet formation

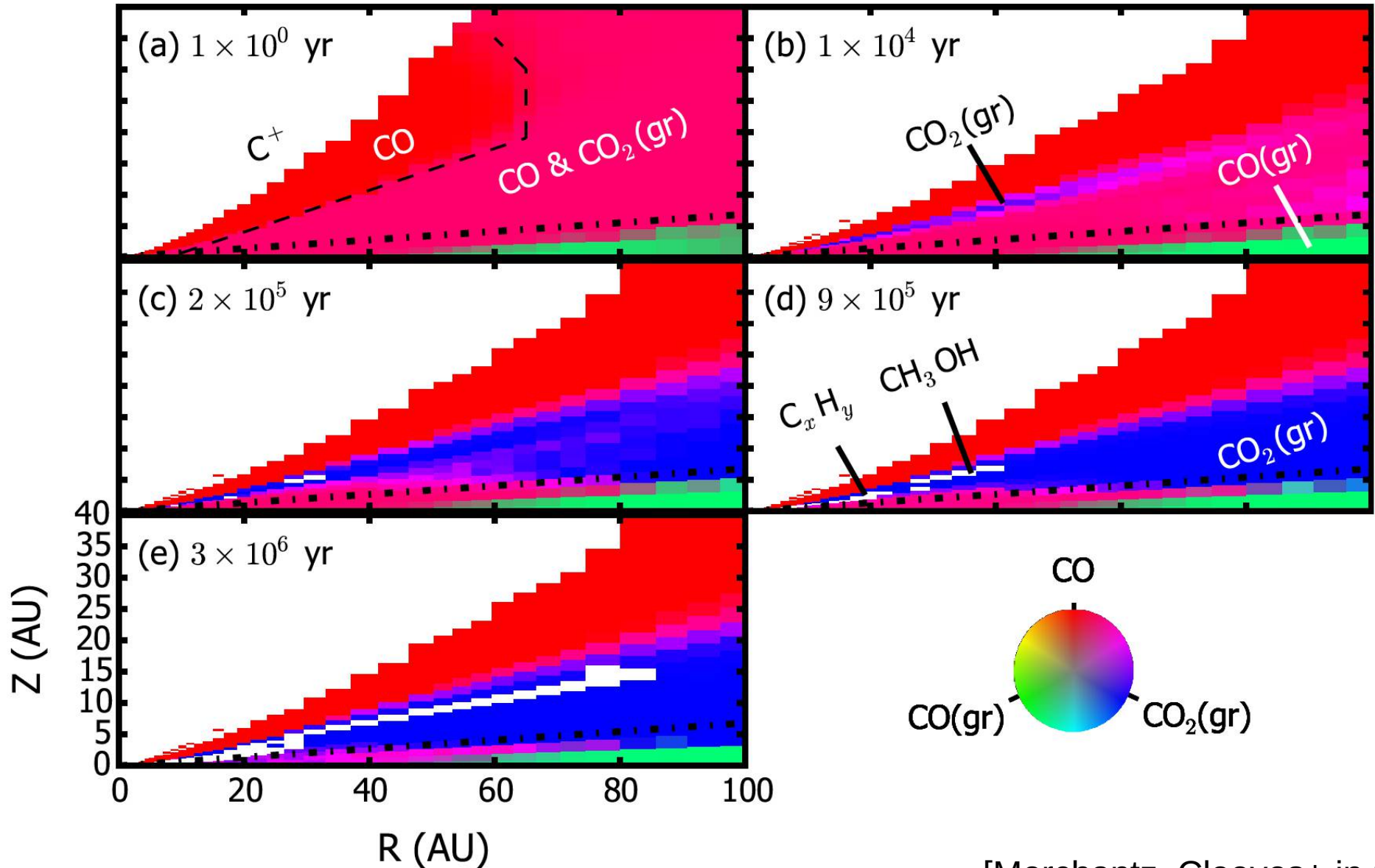


# Planet formation and disk snow lines



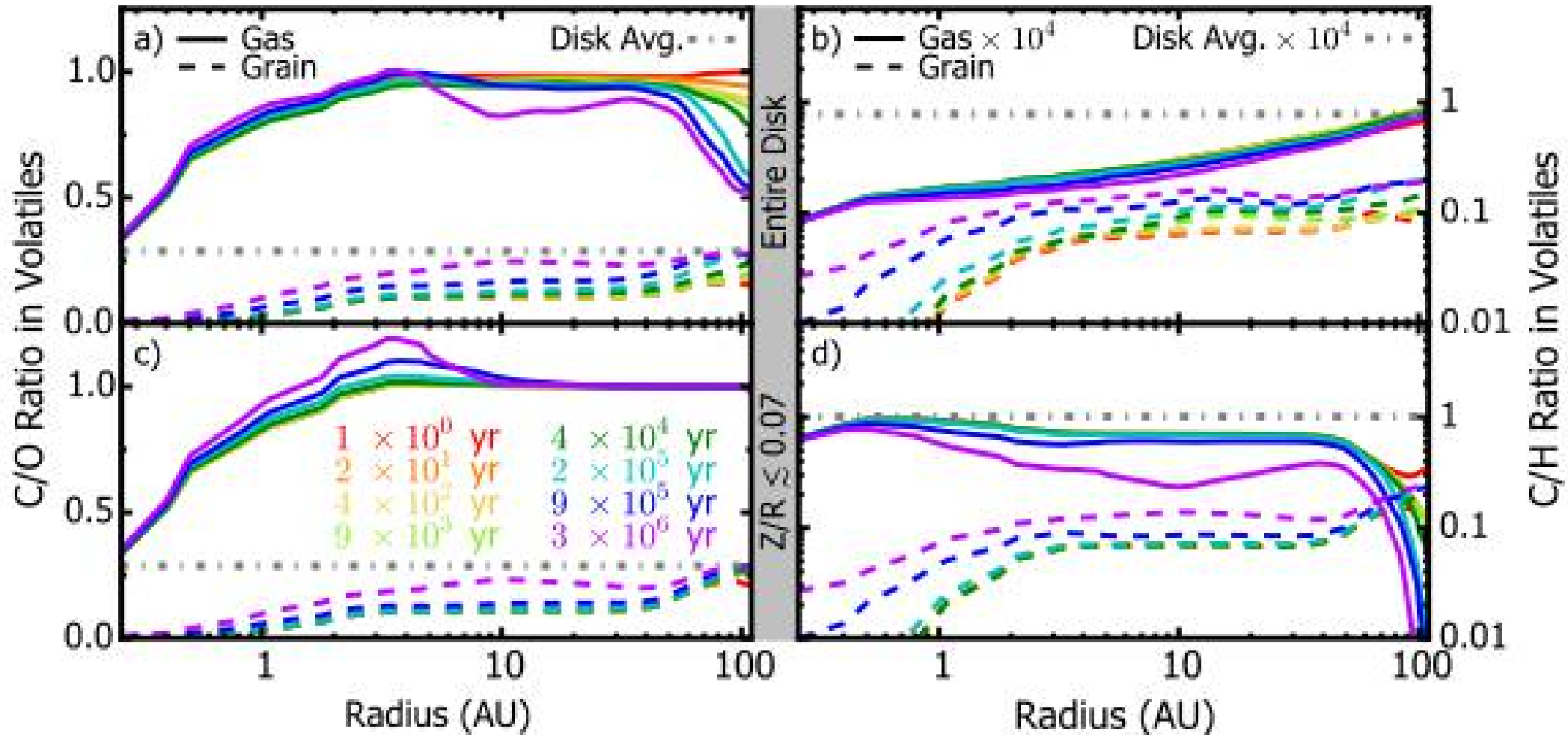


# An evolving chemistry: an evolving gas-ice interaction

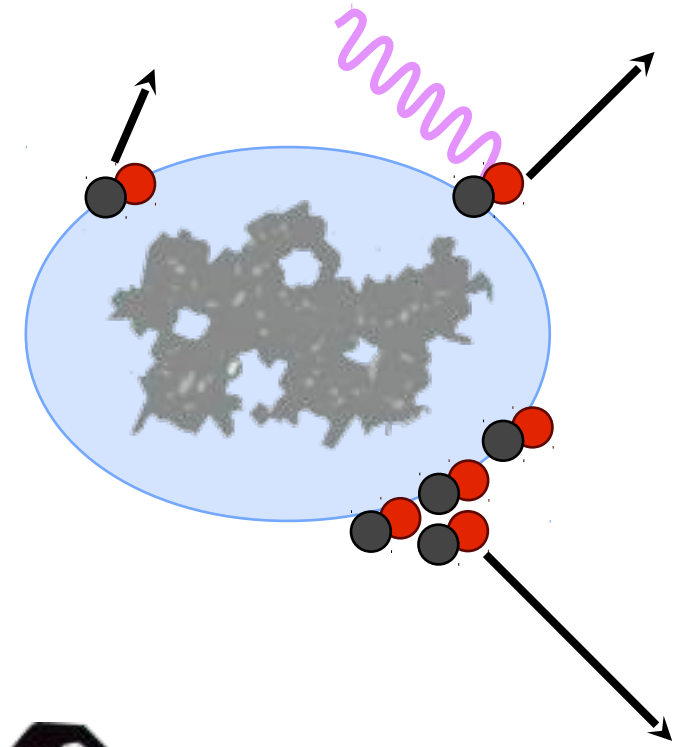
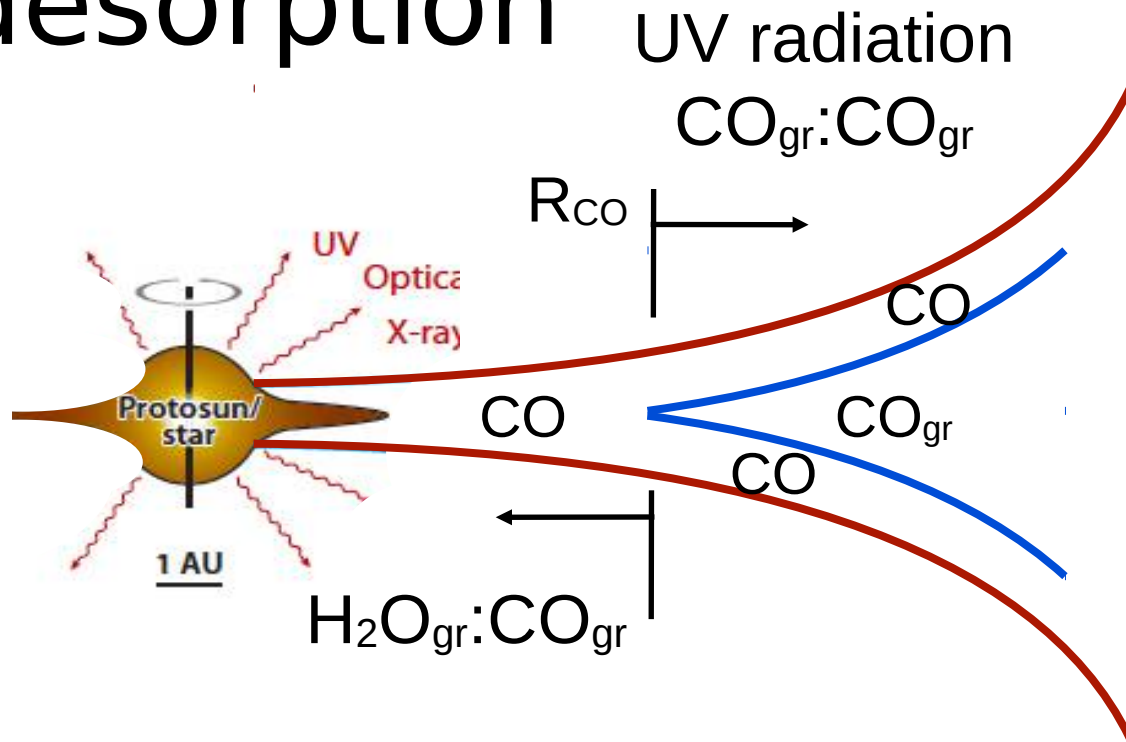


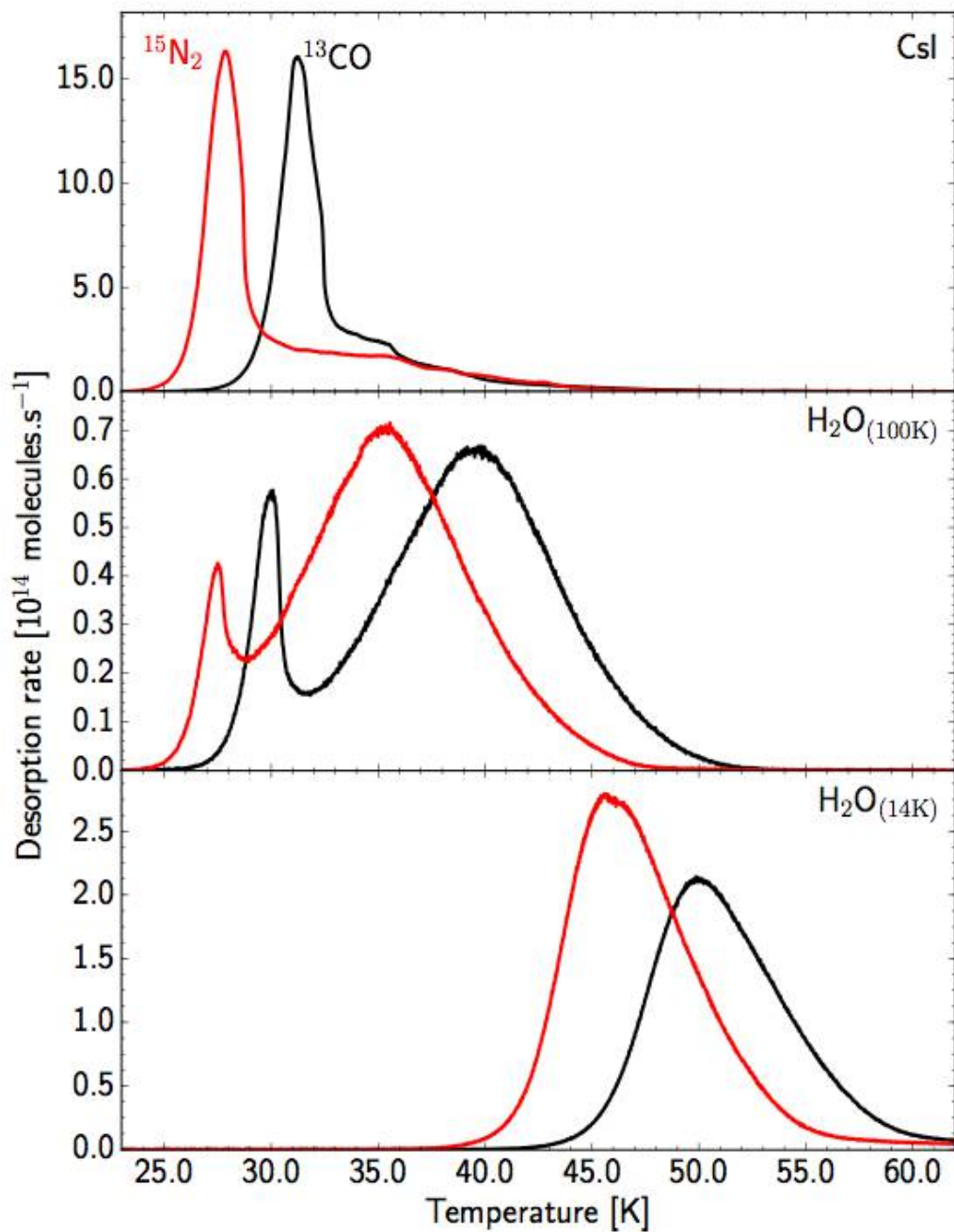
[Merchantz, Cleeves+ in prep.]

# C/O in a chemically evolving disk



# Setting the CO snowline location: binding energies and non-thermal desorption



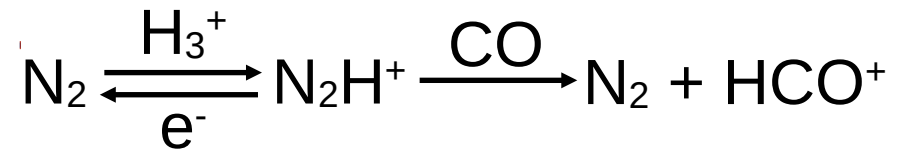
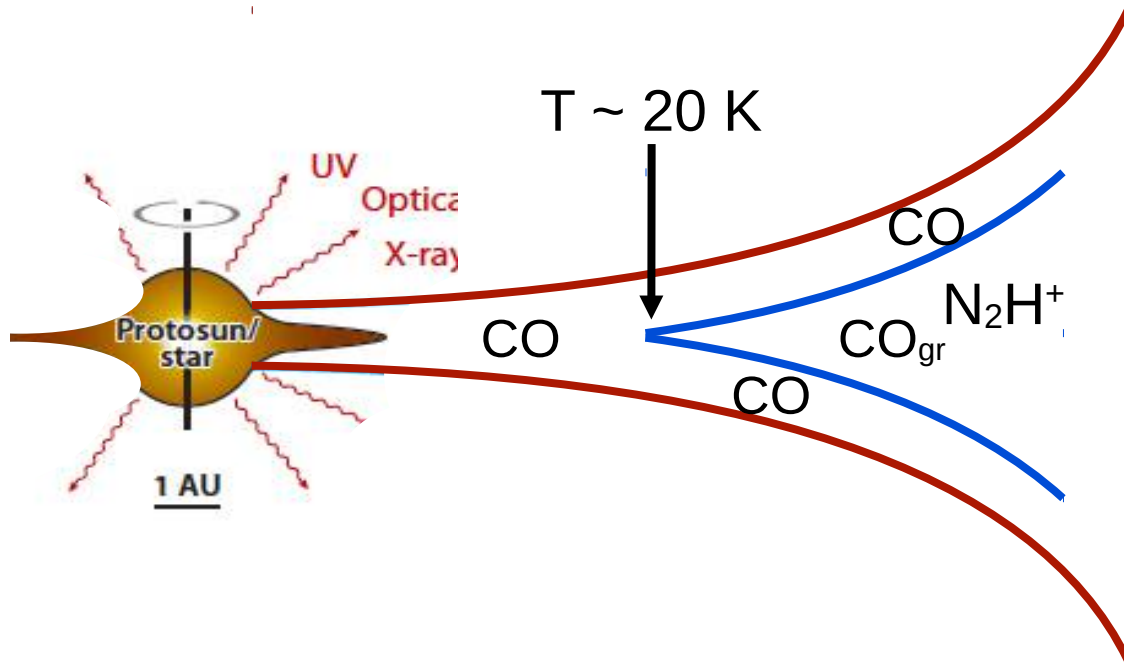


← CO:CO desorbs ~30 K,  
N<sub>2</sub>:N<sub>2</sub> ~27 K

CO and N<sub>2</sub>  
desorption  
energies can vary  
by a factor of 2

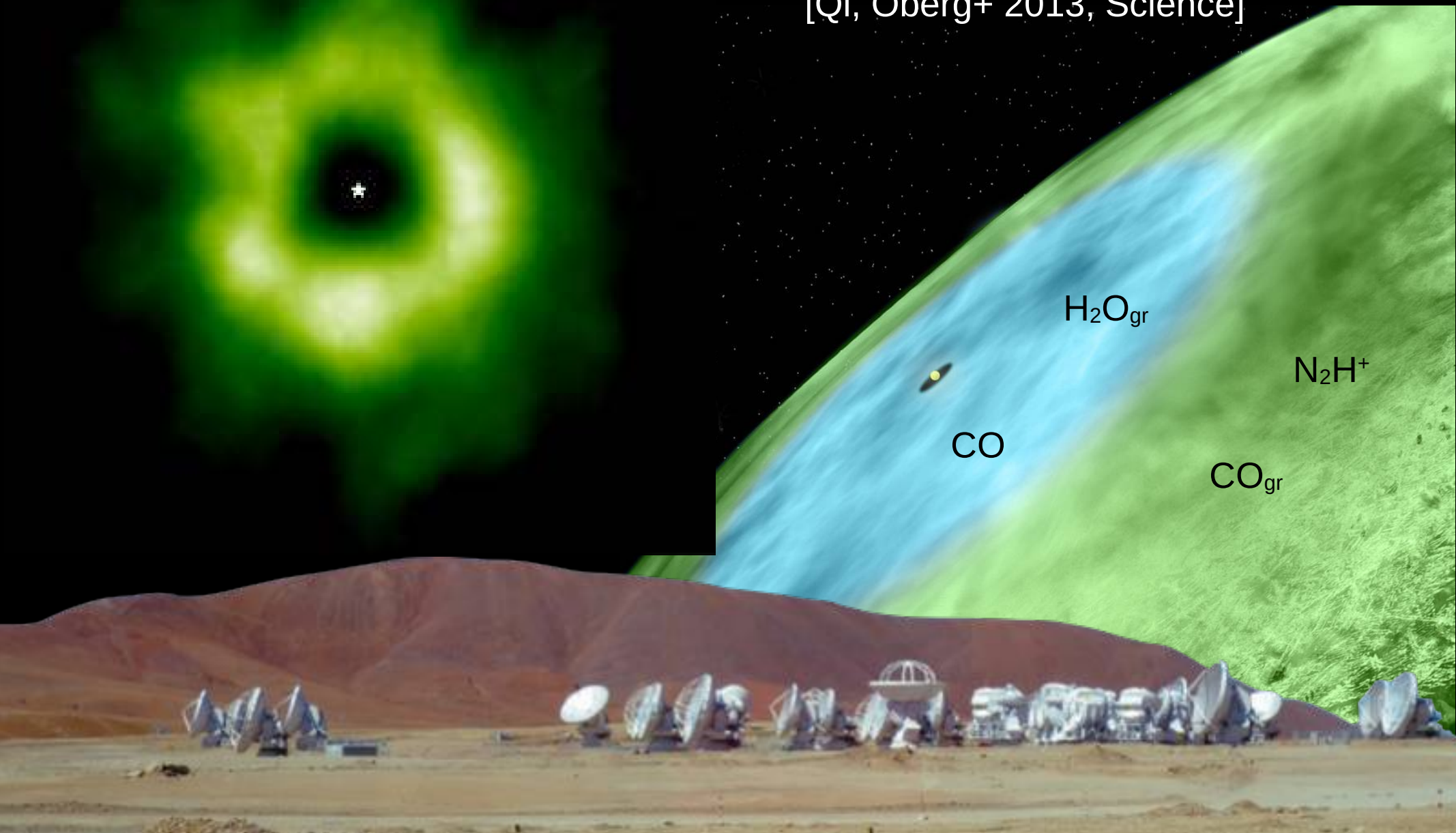
← CO:H<sub>2</sub>O desorbs ~50  
K, N<sub>2</sub>:H<sub>2</sub>O ~45 K

# Imaging the CO snowline in a disk

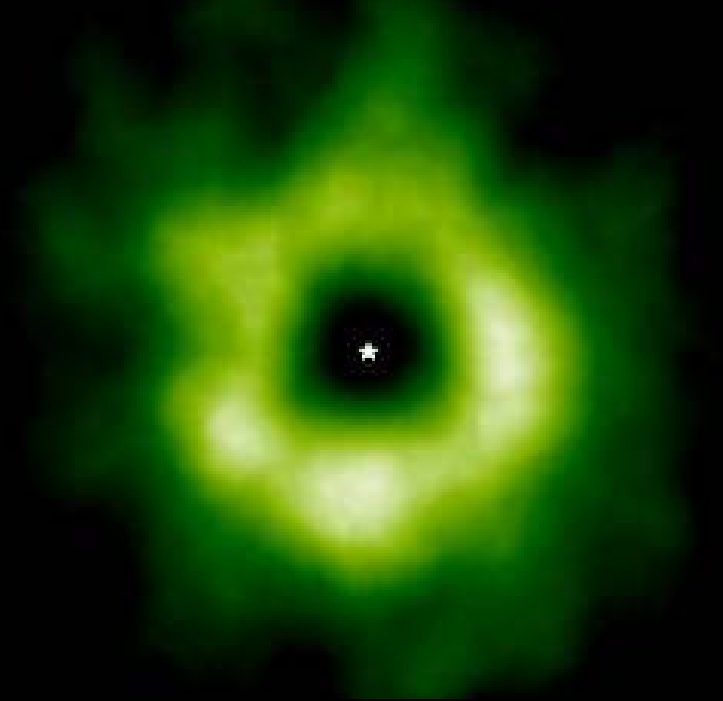


# A chemical image of the CO snowline using $\text{N}_2\text{H}^+$

[Qi, Öberg+ 2013, Science]

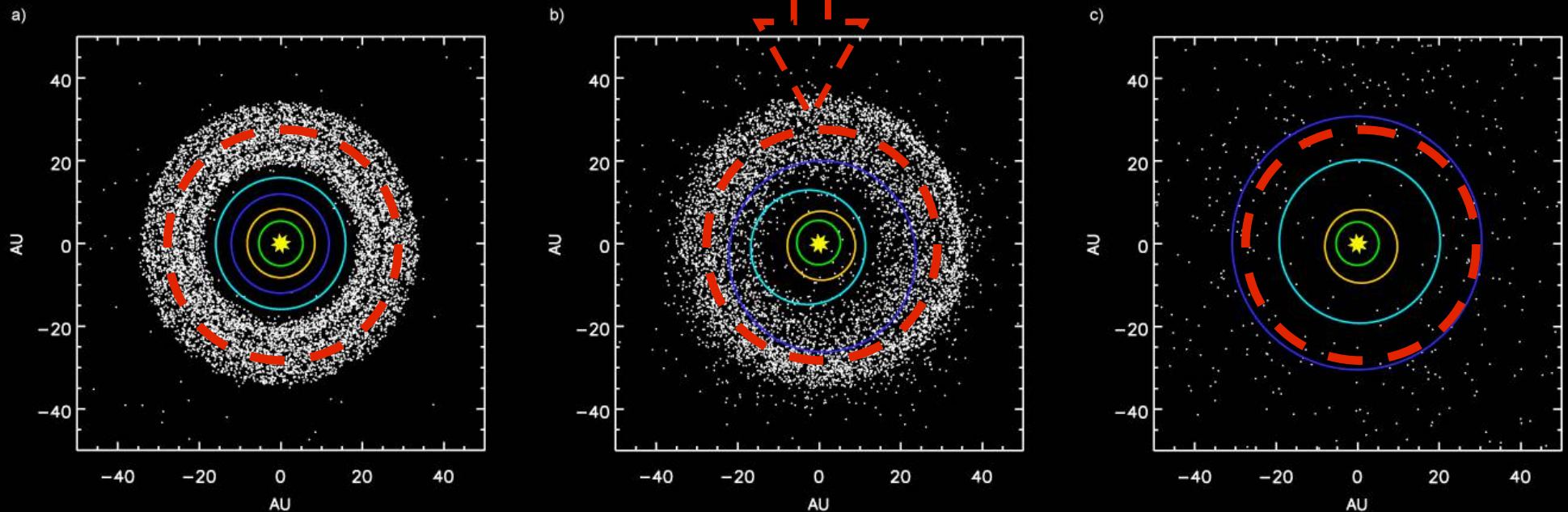


# CO snow line radius implications for the Solar System



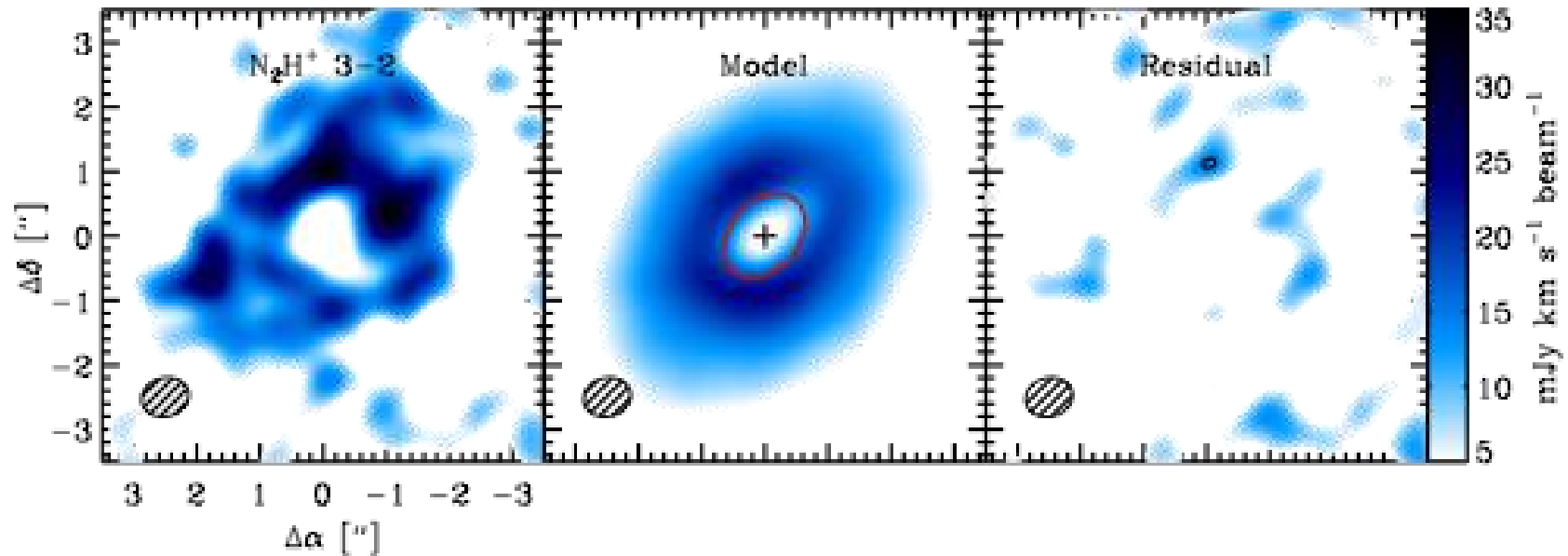
[Tsiganis+ 2005, Gomes+ 2005]

## CO snow line



CO snow line is outside of Ice Giant formation zone according to Nice model. Some comets and KBOs should be CO rich.

# CO snowlines in other disks?



[Qi, Öberg+ ApJ subm.]

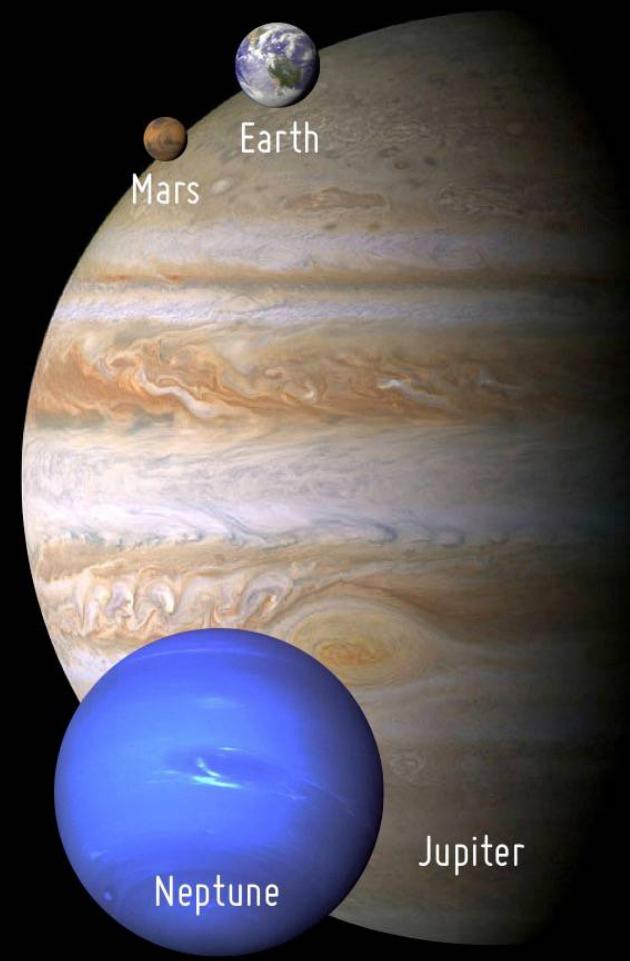
CO snowline in HD 163296 is at  $\sim 90$  AU, corresponding to CO freeze-out at 25 K (cf 18 K in TW Hya)





# Current Potentially Habitable Exoplanets

Ranked in Order of Similarity to Earth

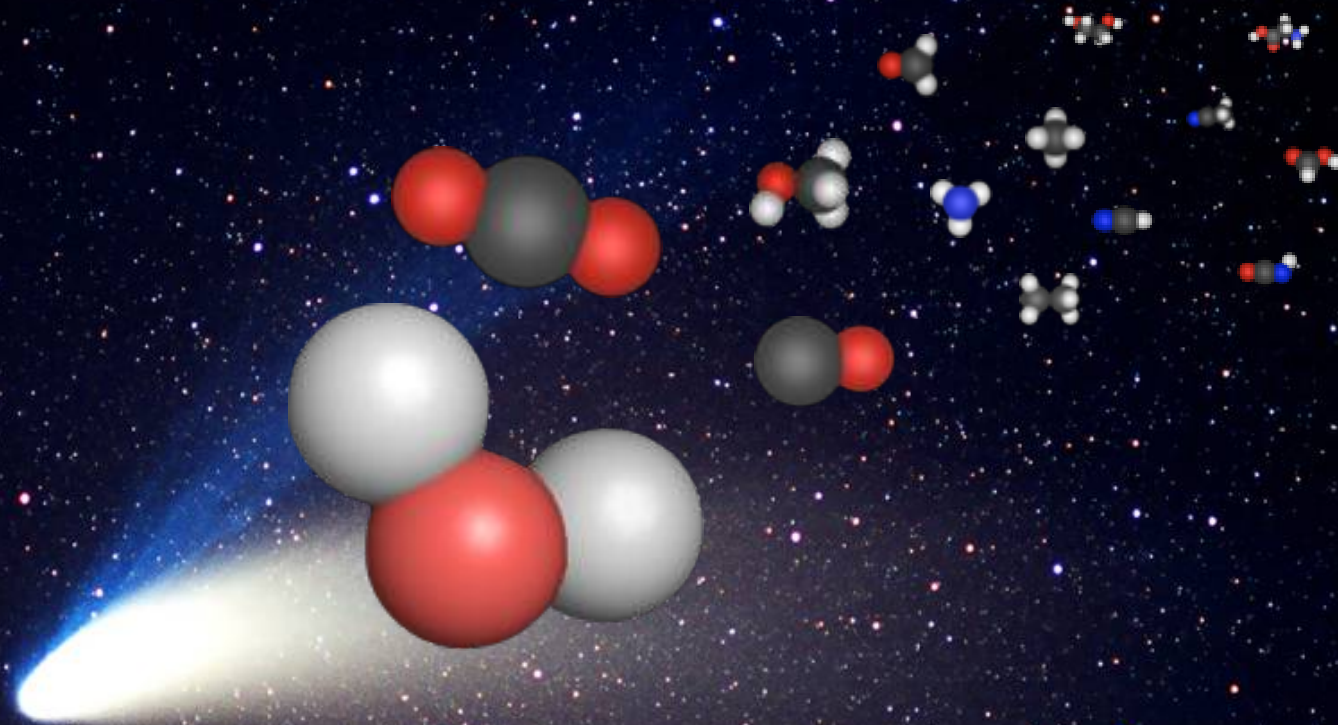


\*planet candidates

CREDIT: PHL @ UPR Arcibo (phl.upr.edu) March 4, 2014

Habitability beyond rocky surfaces and liquid water temperatures... Chemical habitability: access to water, reactive organic molecules... How common is access to the ingredients of life on 'habitable planets'?

# Comet Compositions



# Making a chemically habitable planet

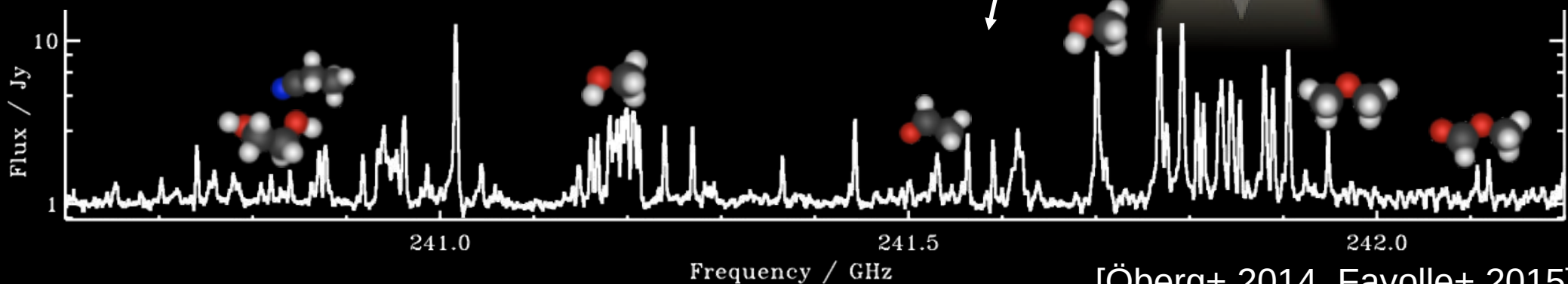
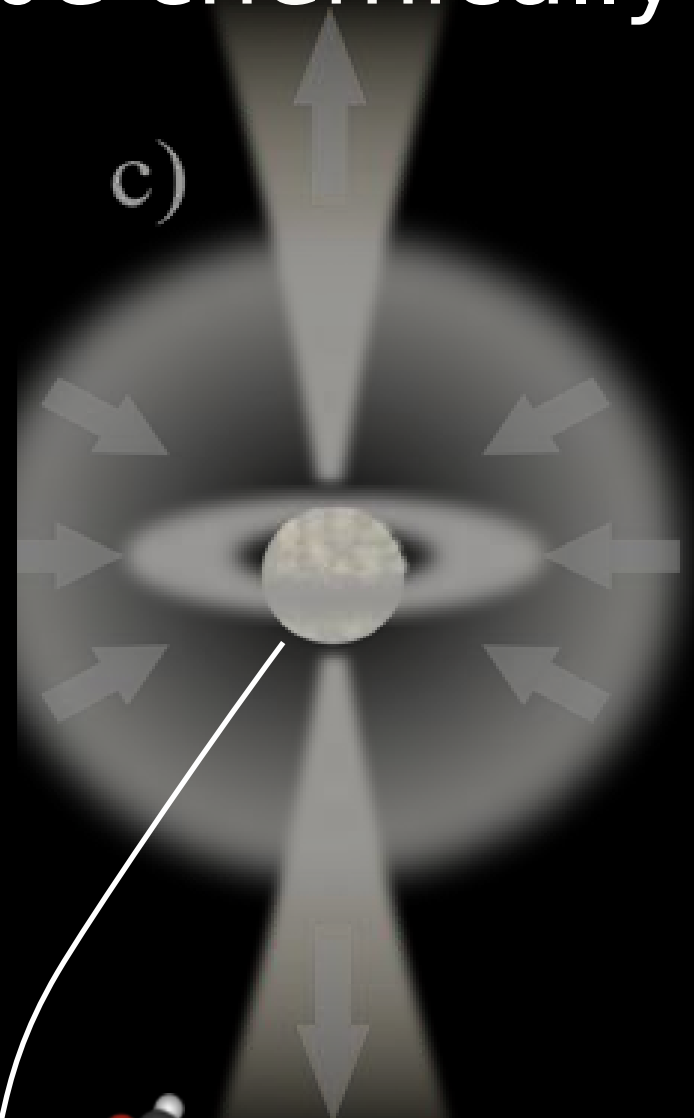


What is the distribution of complex organics during planet formation?



# Massive protostars can be chemically very rich

c)



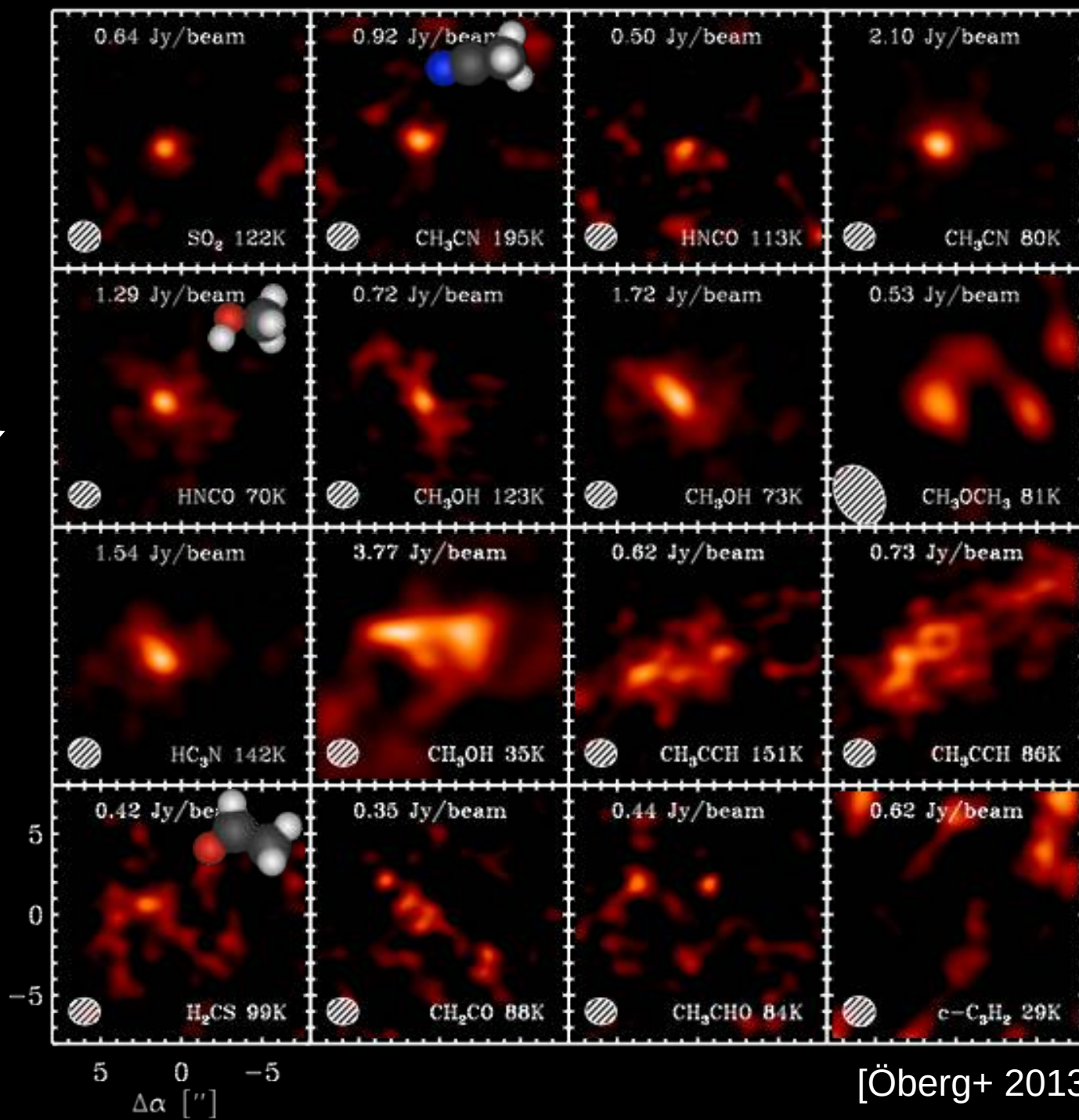
[Öberg+ 2014, Fayolle+ 2015]

# Complex Organic Chemical Images

Compact, unresolved emission,  
e.g. hot  $\text{CH}_3\text{CN}$

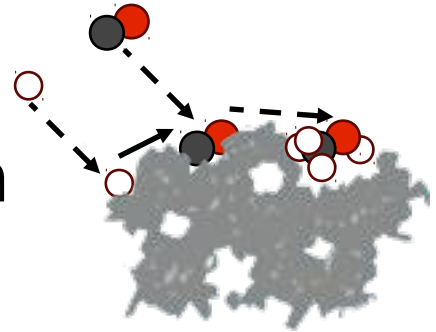
Resolved emission,  
e.g.  $\text{CH}_3\text{OH}$

Resolved out emission,  
e.g.  $\text{CH}_3\text{CHO}$

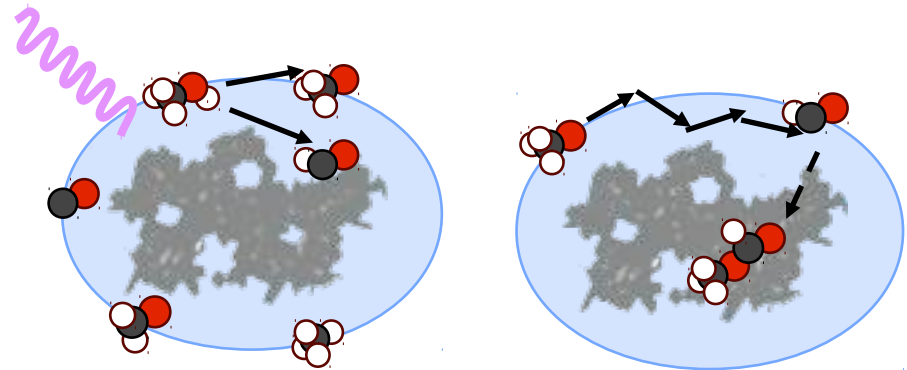


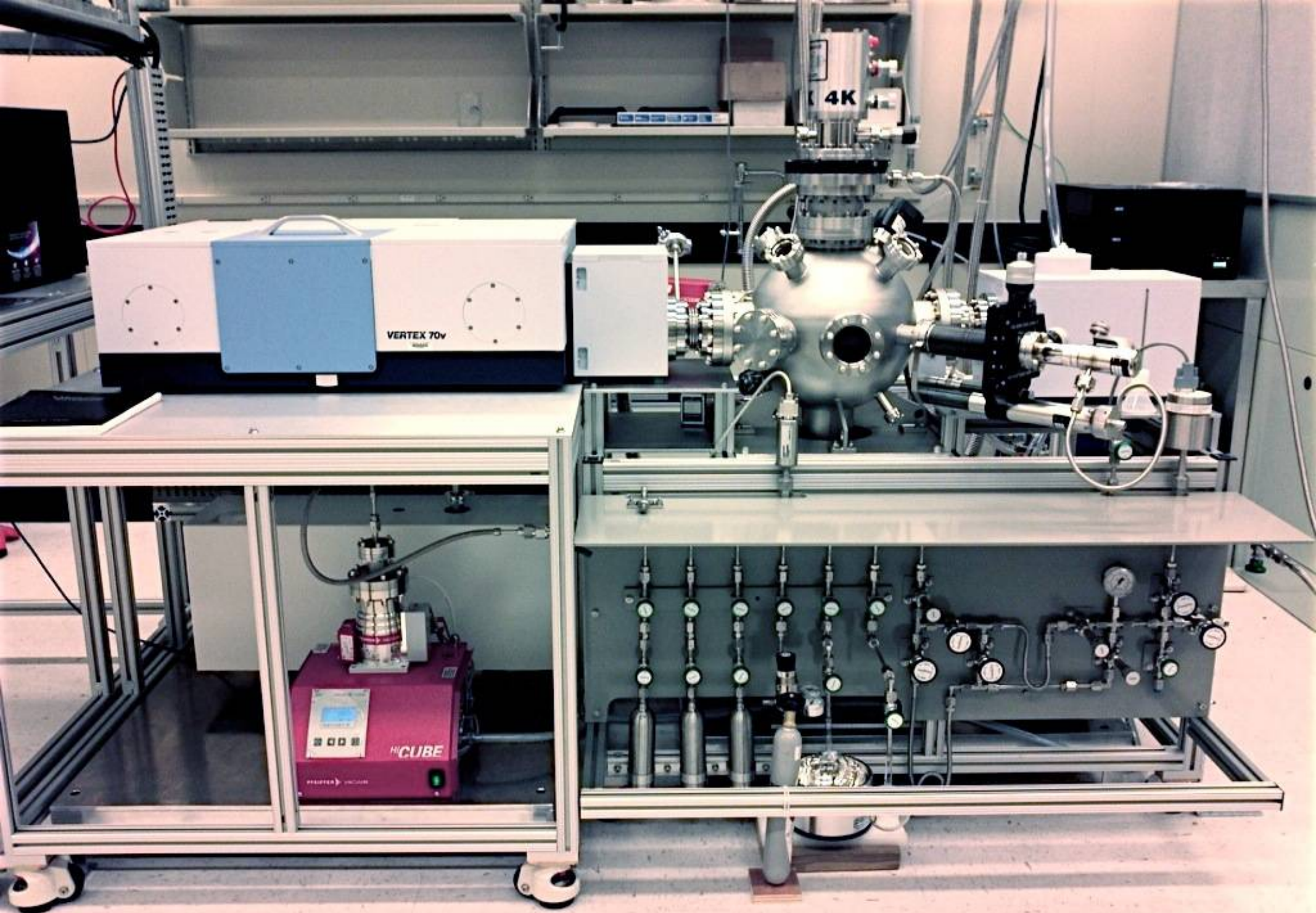
# Formation of complex organics in space

1. Atom and molecule accretion onto grains, accompanied with atom addition reactions



2. Ice photochemistry: ice dissociation, radical diffusion + recombination





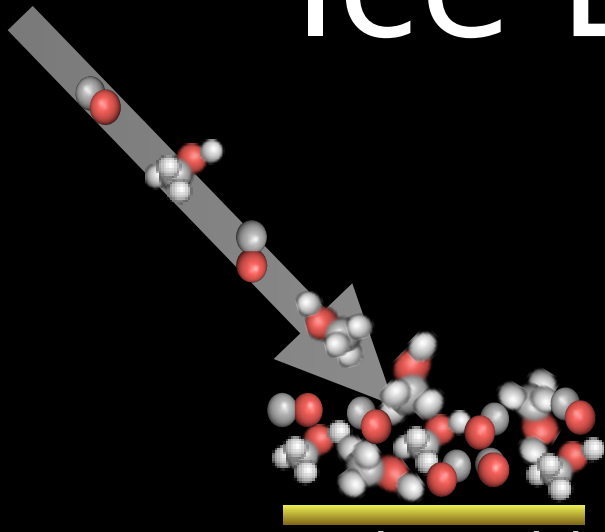
VERTEX 70v

4K

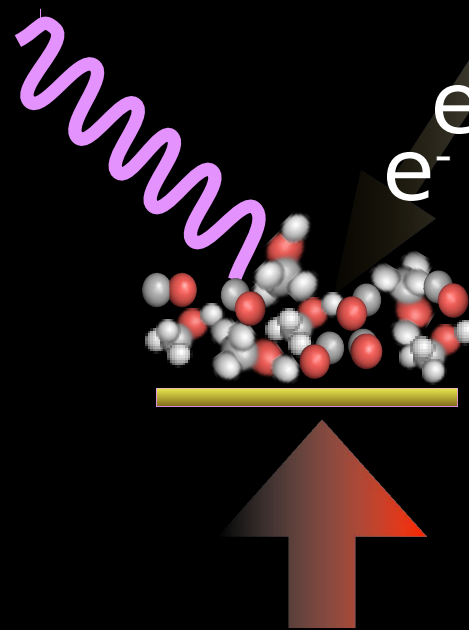
Hi-CUBE



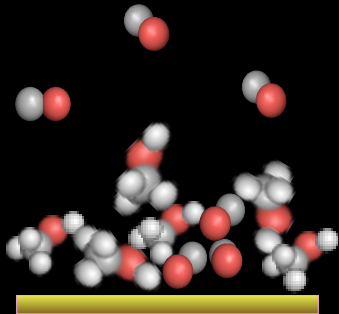
# Ice Experiments



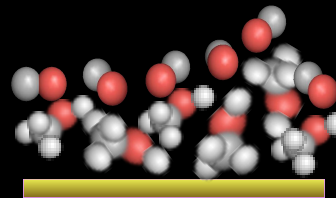
1. Ice deposition:  
can regulate ice  
composition, porosity,  
thickness



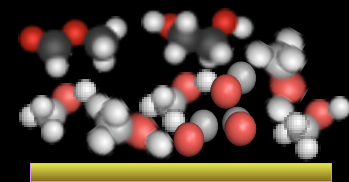
2. Ice manipulation:  
Heat, UV, electrons,  
X-rays  
Continuous and  
pulses, broad-band  
and frequency  
resolved



3a. Ice desorption:  
Thermal and non-thermal

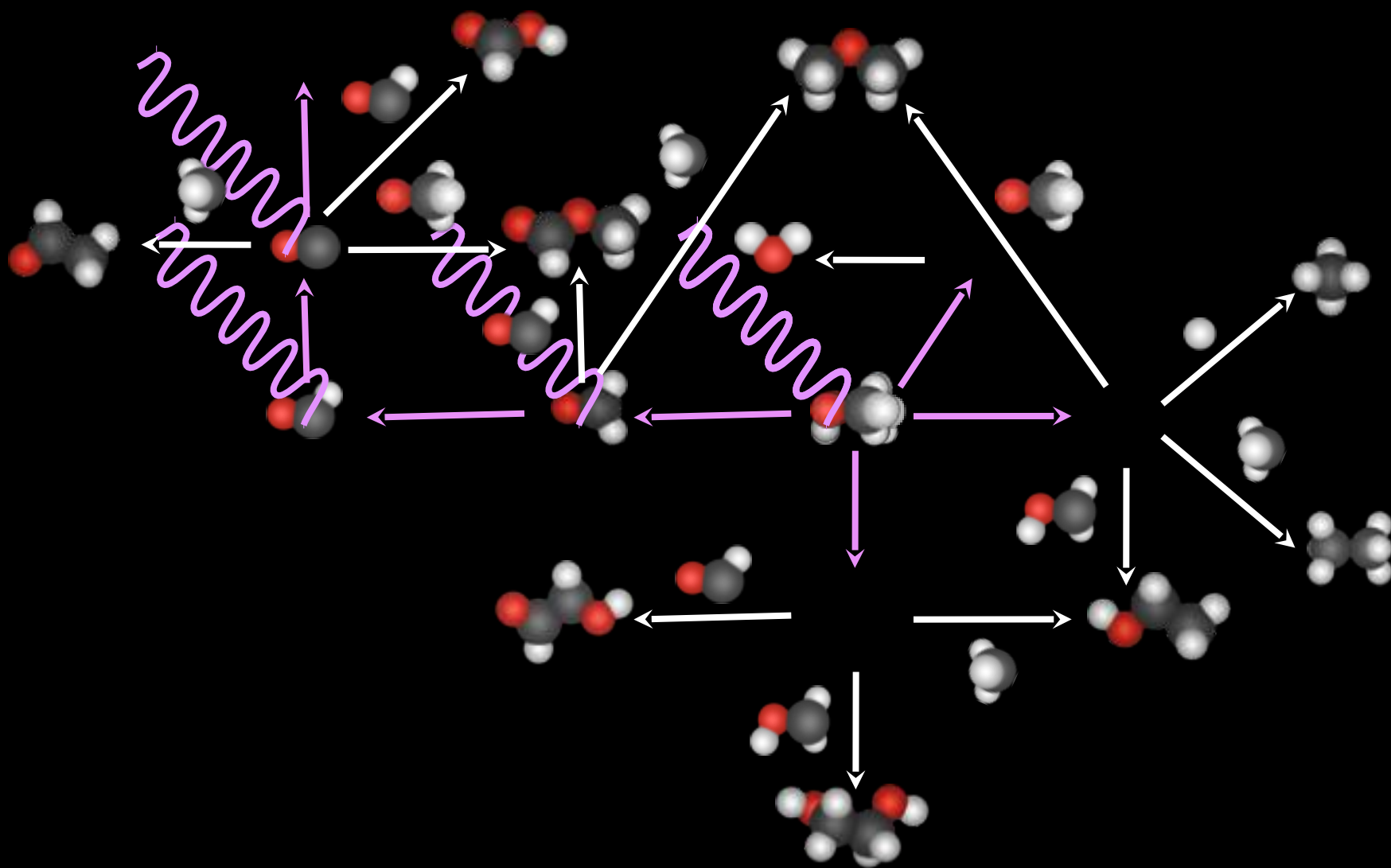


3b. Ice diffusion



3c. Ice chemistry

# CH<sub>3</sub>OH ice is a source of chemical complexity



# Complex organics in space

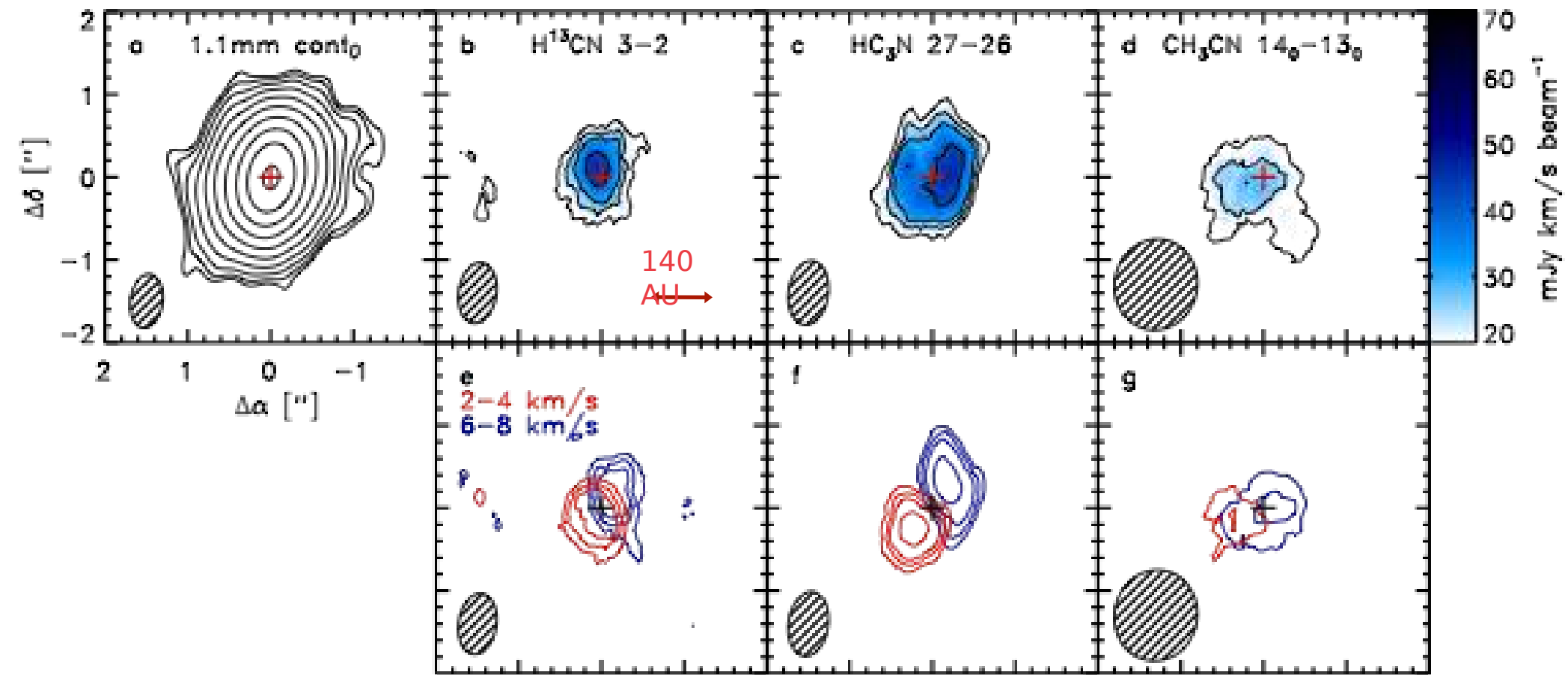
Complex organics are present in all kinds of dense interstellar environments from Hot Cores to Cold Prestellar Cores

The spatial patterns and compositional trends across samples support an ice origin (though many questions remain).

Chemical richness measured as COM/CH<sub>3</sub>OH varies by orders of magnitudes between sources.

What is the distribution of complex organics during planet formation?





ALMA sees first complex organic molecule in a protoplanetary disk!



# Comet-disk comparison

HCN / HC<sub>3</sub>N / CH<sub>3</sub>CN

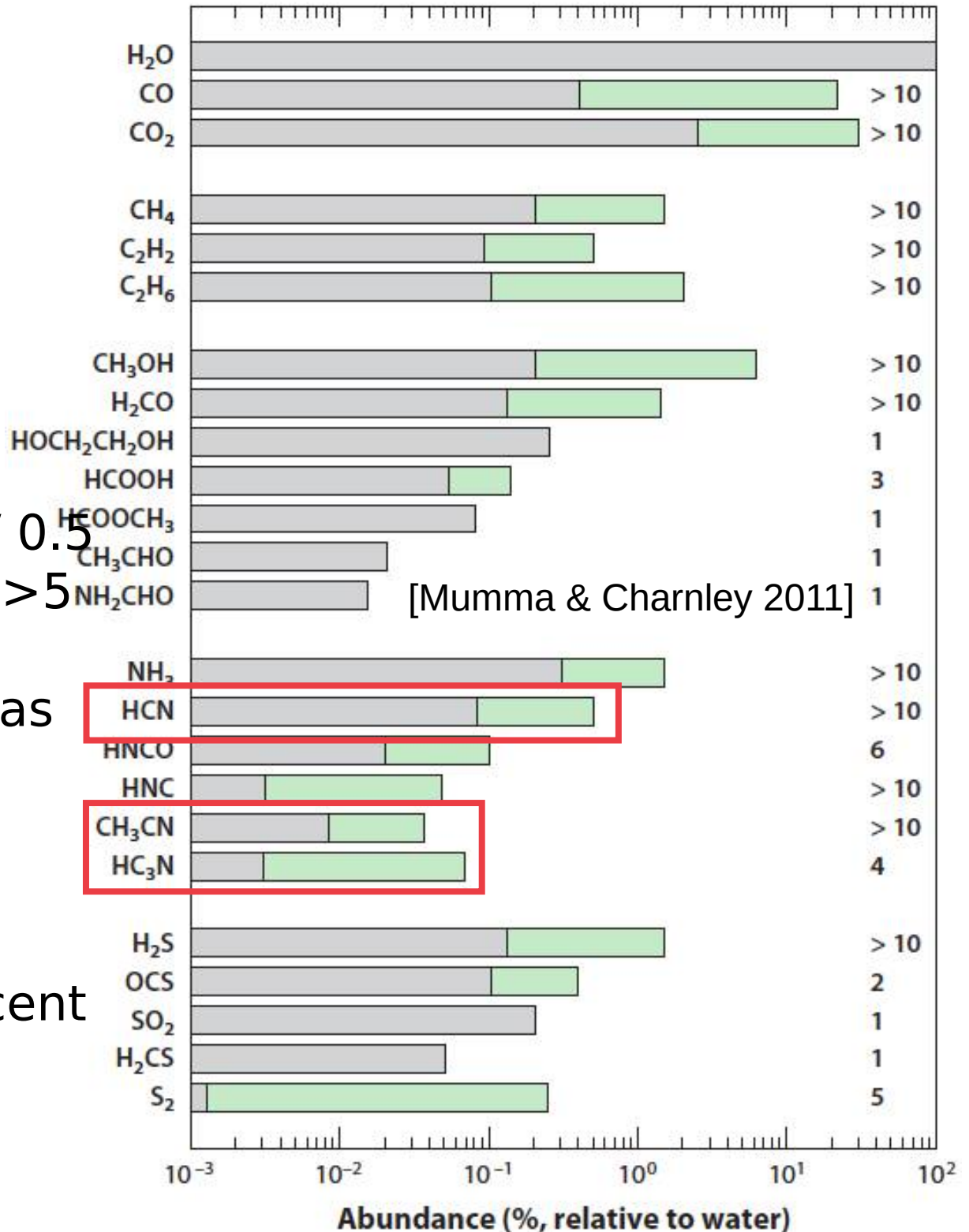
Comets: 10 / 1 / 1

Disk gas 30 AU: ~10 / 4 / 0.5

Disk ice 30 AU: ~10 / 4 / >5

The MWC 480 disk looks as comet in its cyanide composition

The prebiotic conditions characteristic of the nascent Earth may be common?



# The Chemistry of planet formation

The efficiency of planet formation, the final composition of nascent planets, and the access to water and volatile organics (chemical habitability) at planet surfaces are regulated by chemistry in protoplanetary disks

The study of disk chemistry reveals:

- the location of snowlines (through  $\text{N}_2\text{H}^+$ )
- the distribution of complex organic chemistry (discovery of methyl cyanide in a disk)

Through laboratory experiments and interstellar COM observations we can uncover the processes that govern these distributions

